

Question 33 (4 marks)

A student adds 1.17 g of  $\text{Al}(\text{OH})_3(s)$  to 0.500 L of  $0.100 \text{ mol L}^{-1} \text{HCl}(aq)$ .

4

Calculate the pH of the resulting solution. Assume that the volume of the resulting solution is 0.500 L.

① Write a balanced chemical equation



② Find no. of moles of  $\text{OH}^-$  and  $\text{H}^+$

$$n(\text{Al}(\text{OH})_3) = \frac{m}{M_r} = \frac{1.17}{26.98 + 3(16 + 1.008)} = 0.01499 \dots \text{mol}$$

$$n(\text{OH}^-) = 3 \times n(\text{Al}(\text{OH})_3) = 0.04499 \dots \text{mol}$$

$$n(\text{HCl}) = c \times V = 0.5 \times 0.1 = 0.05 \text{ mol}$$

$$n(\text{H}^+) = n(\text{HCl}) = 0.05 \text{ mol}$$

③ Determine which reagent is in excess and how much

As  $n(\text{H}^+) > n(\text{OH}^-)$   $\therefore \text{H}^+$  is in excess

$$n(\text{H}^+)_{\text{excess}} = 0.05 - 0.04499 \dots = 0.00500 \dots \text{mol}$$

④ Calculate the  $[\text{H}^+]$  and thus find pH.

$$[\text{H}^+] = \frac{n}{V} = \frac{0.00500 \dots}{0.500} = 0.0100 \dots \text{mol L}^{-1}$$

$$\text{pH} = -\log_{10} [\text{H}^+] = 2.000 \text{ (3 dp)}$$

Using d.p. 0015 for pH

Please turn over

# Hsc Chemistry 6 User Guide

**Pierre Steinmetz**



## Hsc Chemistry 6 User Guide:

*Excel HSC Chemistry* Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp  
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**T.T. Chen Honorary Symposium on Hydrometallurgy, Electrometallurgy and Materials Characterization** Shijie Wang, J. E. Dutrizac, Michael L. Free, James Y. Hwang, Daniel Kim, 2012-05-09 Proceedings

of a symposium sponsored by the Hydrometallurgy and Electrometallurgy Committee and the Materials Characterization Committee of the Extraction and Processing Division of TMS The Minerals Metals Materials Society Held during the TMS 2012 Annual Meeting Exhibition Orlando Florida USA March 11 15 2012      **Excel HSC Biology** Diane Alford, Jennifer Hill, 2008      **Treatise on Process Metallurgy** Alexander McLean, Roderick Guthrie, Sridhar Seetharaman, H. Y. Sohn, 2024-08-23 Treatise on Process Metallurgy Volume Four Industrial Production provides academics with the fundamentals of the manufacturing of metallic materials from raw materials into finished parts or products In these fully updated volumes coverage is expanded into four volumes including Process Fundamentals encompassing process fundamentals structure and properties of matter thermodynamic aspects of process metallurgy and rate phenomena in process metallurgy Processing Phenomena encompassing interfacial phenomena in high temperature metallurgy metallurgical process phenomena and metallurgical process technology Metallurgical Processes encompassing mineral processing aqueous processing electrochemical material and energy processes and iron and steel technology non ferrous process principles and production technologies and more The work distills the combined academic experience from the principal editor and the multidisciplinary four member editorial board Provides the entire breadth of process metallurgy in a single work Includes in depth knowledge in all key areas of process metallurgy Approaches the topic from an interdisciplinary perspective providing broad range coverage on topics      **Excel HSC Chemistry** C. M. Roebuck, 2003      Excel HSC Survival Guide Pascal Press Staff, 1996      High Temperature Corrosion and Protection of Materials 6 Pierre Steinmetz, 2004      **HSC Mathematics Revision & Exam Workbook** Lyn Baker, 2004 This book has been specifically designed to help Year 12 students to roughly revise all topics in the HSC Mathematics course and prepare for class assessments trial HSC and HSC exams Together with the Year 11 Preliminary Revision Exam Workbook the whole senior Mathematics course is covered The book includes topics covering the complete HSC Mathematics course 200 pages of practice exercises with topic tests for all chapters cross references to relevant pages in the HSC Mathematics study guide topic tests for all chapters two sample examination papers answers to all questions      Handbook of Continuous Crystallization Nima Yazdanpanah, Zoltan K Nagy, 2020-02-04 Continuous crystallization is an area of intense research with particular respect to the pharmaceutical industry and fine chemicals Improvements in continuous crystallization technologies offer chemical industries significant financial gains through reduced expenditure and operational costs and consistent product quality Written by well known leaders in the field The Handbook of Continuous Crystallization presents fundamental and applied knowledge with attention paid to application and scaling up and the burgeoning area of process intensification Beginning with concepts around crystallization techniques and control strategies the reader will learn about experimental methods and computational tools Case studies spanning fine and bulk chemicals the pharmaceutical industry and employing new mathematical tools put theory into context      **Excel Preliminary Information Processes and Technology** Mary

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