

Acids and bases

Chapter outline

- Recall the definitions of Brønsted–Lowry acids and bases and deduce these in a chemical reaction, including the identification of conjugate acid–base pairs.
- Recall the terms ‘amphoteric’ and ‘amphiprotic’.
- Describe the reactions of acids with metals, metal oxides, metal hydroxides, hydrogen carbonates and carbonates.
- Recall that pH is a measure of the concentration of aqueous hydrogen ions and solve problems involving pH and the concentration of hydrogen ions and hydroxide ions.
- Explain the difference between strong and weak acids and bases and describe experiments that could be used to distinguish these.
- Recall what is meant by acid deposition and explain some of the problems associated with acid deposition as well as pre- and post-combustion methods of reducing sulfur oxide emissions.
- Recall the definitions of Lewis acids and bases and recognise these in a chemical reaction. **HL**
- Deduce the dissociation constant for weak acids and bases. **HL**
- Sketch and explain the important features of titration curves including the composition and action of a buffer and use these curves to deduce the pK_a or pK_b value of a weak acid or base. **HL**
- Explain how an acid–base indicator works and select a suitable indicator for a titration. **HL**

KEY TERMS

Acid deposition: A more general term than acid rain; it refers to any process in which acidic substances leave the atmosphere and are deposited on the Earth’s surface.

Brønsted–Lowry acid: A proton (H^+) donor.

Brønsted–Lowry base: A proton (H^+) acceptor.

pH: A measure of the concentration of hydrogen ions in an aqueous solution.
 $pH = -\log_{10}[H^+]$

Strong acid/base: An acid/base that is completely dissociated into its ions in aqueous solution.

Weak acid/base: An acid/base that is only partially dissociated into its ions in aqueous solution.

Guided Reading And Study Workbook Chapter 20 Acids Bases

R Barnett



Guided Reading And Study Workbook Chapter 20 Acids Bases:

Embracing the Melody of Expression: An Psychological Symphony within **Guided Reading And Study Workbook Chapter 20 Acids Bases**

In some sort of consumed by screens and the ceaseless chatter of immediate conversation, the melodic beauty and mental symphony created by the prepared word often disappear in to the background, eclipsed by the relentless noise and disruptions that permeate our lives. Nevertheless, nestled within the pages of **Guided Reading And Study Workbook Chapter 20 Acids Bases** an enchanting fictional value brimming with organic feelings, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, this charming masterpiece conducts visitors on an emotional journey, skillfully unraveling the concealed tunes and profound influence resonating within each carefully constructed phrase. Within the depths of this poignant assessment, we can examine the book is main harmonies, analyze its enthralling publishing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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