Practice Problems: Double Displaceme

1. ___ $Fe(s)$ + ___ $H_2SO_4(aq)$ → ___ $Fe_2(SO_4)_3(aq)$ + ___

2. ___ $NaC_2H_3O_2(aq)$ + ___ $H_2CO_3(aq)$ → ___ $Na_2CO_3(aq$

3. ___ $(NH_4)_3PO_4(aq)$ + ___ $Pb(NO_3)_4(aq)$ → ___ $Pb_3(PO_4$

4. ___ $C_4H_{10}(g)$ + ___ $O_2(g)$ → ___ $CO_2(g)$ + ___ $H_2O(g)$

5. ___ $H_3PO_4(aq)$ + ___ $HCl(aq)$ → ___ $PCl_5(s)$ + ___ H_2C

6. ___ $Hg(OH)_2(s)$ + ___ $H_3AsO_4(aq)$ → ___ $Hg_3(AsO_4)_2(s$

7. ___ $HClO_4(aq)$ + ___ $P_4O_{10}(s)$ → ___ $H_3PO_4(aq)$ +

Guided Practice Problem Chemistry

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