

Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
 - Physical changes result in changes of appearance, but not composition.
 - Melting, evaporation, freezing, etc.
 - Chemical changes result in changes of composition of matter. (Chemical reaction)
 - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
 - Physical properties can be described by observing the matter without a change in the composition.
 - Color, odor, density, temperature, etc.
 - Chemical properties can only be detected by altering the composition.
 - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
 - Atoms are the smallest unit of a given element that carry the properties of that element.
 - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
 - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
 - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
 - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
 - Correctness
 - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
 - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
 - All nonzero digits ARE significant.
 - Zeros between nonzero digits ARE significant.
 - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
 - Zeros at the end of a number to the right of a decimal point ARE significant.
 - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

General Chemistry Final Exam Study Guide

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General Chemistry Final Exam Study Guide:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **General Chemistry Final Exam Study Guide**

In a world inundated with monitors and the cacophony of instantaneous transmission, the profound energy and emotional resonance of verbal artistry often disappear into obscurity, eclipsed by the constant barrage of noise and distractions. However, situated within the musical pages of **General Chemistry Final Exam Study Guide**, a charming work of fictional beauty that pulses with organic thoughts, lies an wonderful trip waiting to be embarked upon. Penned by a virtuoso wordsmith, that enchanting opus manuals visitors on an emotional odyssey, softly revealing the latent possible and profound affect embedded within the intricate web of language. Within the heart-wrenching expanse of this evocative evaluation, we can embark upon an introspective exploration of the book is central styles, dissect their captivating writing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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