

Which of these are **true** and which are **false** about acid-base reactions?

- In general, an acid-base reaction will be favored when a weaker acid and a weaker base combine to form a stronger acid and a stronger base T **F**
Other way around: will be favored when a stronger acid and a stronger base combine to give a weaker acid and a weaker base.
- Generally, strong acids have weak conjugate bases T **F**
True - and weak acids have strong conjugate bases
- Strong acids tend to have high pK_a values T **F**
Strong acids have low (strongly negative) pK_a values - e.g. HCl has a pK_a of -8
- The greater the acidity of a molecule, the weaker a base it is T **F**
*Not quite correct. The stronger the acid, the weaker is its **conjugate** base.*
- The inductive effect of electronegative atoms near a carboxylic acid group increases the acidity of the compound. T **F**
- Any factor which stabilizes negative charge generally tends to decrease the acidity of an acid T **F**
*False - **increases** acidity, not decreases.*
- The pK_a of NH_4 is 10 and the pK_a of H_2O is 14, which means that H_2O is a stronger acid than H_2O T **F**
False - NH_4 is a stronger acid by a factor of 4 pK_a units (about 10^4)
- The acidity of a molecule can be increased by resonance stabilization of its conjugate base. T **F**
- sp -hybridized C-H bonds are more acidic than sp^3 -hybridized C-H bonds since the resulting negative charge is closer to the nucleus T **F**
- Acid-base reactions where a stronger base is formed from a weaker base never occur. T **F**
*They **do** occur and they can be important. E.g. NH_3 can deprotonate H_2O even though it is a weaker base by a factor of 10^4 compared to HO^- . Even a difference in 9-10 pK_a units can be significant for moving some reactions forward.*

General Chemistry Acid Base Practice Problems

Tobias Bleicker



General Chemistry Acid Base Practice Problems:

Adopting the Beat of Phrase: An Mental Symphony within **General Chemistry Acid Base Practice Problems**

In a world used by monitors and the ceaseless chatter of instantaneous conversation, the melodic elegance and psychological symphony produced by the written term frequently disappear into the backdrop, eclipsed by the relentless sound and distractions that permeate our lives. Nevertheless, set within the pages of **General Chemistry Acid Base Practice Problems** an enchanting fictional value full of raw emotions, lies an immersive symphony waiting to be embraced. Constructed by an outstanding musician of language, that captivating masterpiece conducts viewers on a mental trip, well unraveling the concealed tunes and profound affect resonating within each cautiously crafted phrase. Within the depths of this poignant assessment, we shall discover the book is main harmonies, analyze their enthralling publishing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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