Kamer____

Date Pd

Unit 8 Worksheet 3: Molar Concentration

Part A: Describing Molar Concentration

- Sodium chloride was dissolved in water to produce a 1.5M solution. Explain what this concentration tells us about the NaCl solution. How might a chemist use this ratio?
- A 45.3 g sample of potassium nitrate is dissolved in enough water to make 225 mL of solution. Determine the molar concentration of the potassium nitrate.
- Find the molarity of a solution made from 275 g of CuSO₄ dissolved in enough water to make 4.25 L.

 An alcoholic iodine solution ("tincture" of iodine) is prepared by dissolving 5.15 g of iodine crystals in enough alcohol to make a volume of 225 mL. Calculate the molarity of iodine in the solution.

5. What final volume would be needed in order to prepare a 0.25 M NaCl solution from 5.2 g of NaCl (s)?

General Chemistry Unit 1worksheet 4 Molar Concentration

Léo Pomar

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In some sort of taken by monitors and the ceaseless chatter of instant connection, the melodic elegance and mental symphony developed by the prepared term usually disappear in to the back ground, eclipsed by the constant noise and disturbances that permeate our lives. But, situated within the pages of **General Chemistry Unit 1worksheet 4 Molar Concentration** a wonderful fictional treasure overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, this interesting masterpiece conducts visitors on a psychological trip, skillfully unraveling the hidden songs and profound impact resonating within each carefully crafted phrase. Within the depths of the touching assessment, we will explore the book is central harmonies, analyze its enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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