

Experiment 29 Prelaboratory Assignment

Bleach Analysis

Date _____ Lab Sec. _____ Name _____ Desk No. _____

- Potassium iodate, KIO_3 , and the bleach's hypochlorite ion, ClO^- , are oxidizing agents. What do they oxidize in this experiment?
 - Sodium thiosulfate, $\text{Na}_2\text{S}_2\text{O}_3$, is a reducing agent. What does it reduce in this experiment?
- Write the balanced net-ionic equation for the standardization of the sodium thiosulfate solution by combining equations 29.7 and 29.4.
- Experimental Procedure, Part A.1. Describe the procedure for the preparation of 100 mL of a 0.010 M KIO_3 solution, starting with solid KIO_3 (molar mass = 214.02 g/mol).
 - Experimental Procedure, Part B.1. Describe the procedure for the preparation of 250 mL of 0.10 M $\text{Na}_2\text{S}_2\text{O}_3$ starting with solid $\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$ (molar mass = 248.2 g/mol).
- Experimental Procedure, Part B.3. Explain why a minimum, and not a maximum, mass of ~2 g of KI is critical for the standardization of the sodium thiosulfate solution.
- Write the balanced net-ionic equation for the analysis of the "available chlorine" in a sample using sodium thiosulfate as the reducing agent by combining equations 29.2, 29.3, and 29.4.



General Chemistry Pre Lab Manual Answers

Joacim Rocklöv



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**General Chemistry Pre Lab Manual Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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