Basic Transport Phenomena in Biomedical Engineering 4th Edition Fournier Solutions Manual

Chapter I Solutions

Problem 1.1. Use the conversion factors

$$\frac{8.314J}{\text{mol K}} \times \frac{0.23901\text{cal}}{J} = 1.987 \frac{\text{cal}}{\text{mol K}}$$

Problem 1.2. Use Equation 1.2 to convert the temperature to degrees Celsius

$$t \cdot C = \frac{5}{9} (t \cdot F - 32) = \frac{5}{9} (98.6 - 32) = 37 \cdot C$$

Now, use Equation 1.3 to convert the temperature from degrees Celsius to Kelvin

$$TK = t \cdot C + 273.15 = 37 + 273.15 = 310.15 K$$

Problem 1.3. Use the conversion factors

$$2 \times 10^4 \,\mu dynes \times \frac{dynes}{10^6 \,\mu dynes} \times \frac{N}{105 dynes} \times \frac{kN}{10^3 \,N} = 2 \times 10^{-10} \,kN$$

Problem 1.4. Use the conversion factors

$$\frac{27\,\mu m}{hr} \times \frac{10^{-6} \text{ m}}{\mu m} \times \frac{0.0006214 \text{miles}}{m} \times \frac{hr}{60 \text{min}} \times \frac{min}{60 \text{sec}} = 4.66 \times 10^{-12} \frac{\text{miles}}{\text{sec}}$$

Problem 1.5. Use the conversion factors

$$48MW \times \frac{106W}{MW} \times \frac{kg \text{ m}^2}{\text{sec}^3 \text{ W}} \times \frac{10^3 \text{ g}}{\text{kg}} \times \frac{100 \text{cg}}{\text{g}} \times \frac{3600^3 \text{ sec}^3}{\text{hr}^5} \times \frac{100^2 \text{ cm}^2}{\text{m}^2} = 2.24 \times 10^{27} \frac{\text{cgcm}^2}{\text{hr}^3}$$

$$48MW \times \frac{10^6 \text{ W}}{MW} \times \frac{J}{\text{sec W}} \times \frac{kJ}{10^3 \text{ J}} \times \frac{60 \text{sec}}{-\text{min}} = 2.88 \times 10^6 \frac{kJ}{\text{min}}$$

Problem 1.6. Use the conversion factors to get both values into units of mm per second and then compare the two values for the higher value.

Geiger Transport Phenomena Solutions Manual

E. J. Poirier, D. R. Poirier, G. H. Geiger

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Table of Contents Geiger Transport Phenomena Solutions Manual

- 1. Understanding the eBook Geiger Transport Phenomena Solutions Manual
 - The Rise of Digital Reading Geiger Transport Phenomena Solutions Manual
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Geiger Transport Phenomena Solutions Manual
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Geiger Transport Phenomena Solutions Manual
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Geiger Transport Phenomena Solutions Manual
 - Personalized Recommendations
 - Geiger Transport Phenomena Solutions Manual User Reviews and Ratings
 - Geiger Transport Phenomena Solutions Manual and Bestseller Lists
- 5. Accessing Geiger Transport Phenomena Solutions Manual Free and Paid eBooks
 - Geiger Transport Phenomena Solutions Manual Public Domain eBooks
 - Geiger Transport Phenomena Solutions Manual eBook Subscription Services
 - Geiger Transport Phenomena Solutions Manual Budget-Friendly Options
- 6. Navigating Geiger Transport Phenomena Solutions Manual eBook Formats
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 - Geiger Transport Phenomena Solutions Manual Compatibility with Devices
 - Geiger Transport Phenomena Solutions Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Geiger Transport Phenomena Solutions Manual
 - Highlighting and Note-Taking Geiger Transport Phenomena Solutions Manual
 - Interactive Elements Geiger Transport Phenomena Solutions Manual
- 8. Staying Engaged with Geiger Transport Phenomena Solutions Manual

- o Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Geiger Transport Phenomena Solutions Manual
- 9. Balancing eBooks and Physical Books Geiger Transport Phenomena Solutions Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Geiger Transport Phenomena Solutions Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Geiger Transport Phenomena Solutions Manual
 - Setting Reading Goals Geiger Transport Phenomena Solutions Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Geiger Transport Phenomena Solutions Manual
 - Fact-Checking eBook Content of Geiger Transport Phenomena Solutions Manual
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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