

**Fourth Semester B.E. Degree Examination, June-July 2009**  
**Manufacturing Process - II**

Time: 3 hrs.

Max. Marks:100

*Note : Answer any FIVE full questions, selecting atleast  
 TWO questions form each Part – A and Part - B.*

**PART - A**

1.
  - a. With neat sketches, give nomenclature of a single point cutting tool. (07 Marks)
  - b. Derive the conditions  $2\phi + \beta - \gamma = \frac{\pi}{2}$ . Explain its significance with usual notations. (07 Marks)
  - c. List various factors influencing tool – life. Explain any one of them. (06 Marks)
2.
  - a. List various cutting tool materials in the increasing order of their hardness. (06 Marks)
  - b. Explain with the help of neat sketches, heat distribution at tool and workpiece. (06 Marks)
  - c. What are the methods of measurement of temperature of cutting tool tip? Explain any one of them. (08 Marks)
3.
  - a. Differentiate between capstan and turret lathes. (05 Marks)
  - b. With a neat sketch, explain any one of the driving mechanisms used in a shaping machine. (08 Marks)
  - c. Sketch a planning machine indicating major parts. Name any one of the mechanism for quick return movement in a planer. (07 Marks)
4.
  - a. List various drilling machines. (05 Marks)
  - b. Draw neat sketch of a radial drilling machine indicating parts. (06 Marks)
  - c. Sketch and mention the application of the drilling machine operations – counter boring reaming and trepanning. (09 Marks)

**PART - B**

5.
  - a. With sketches, differentiate between up milling and down milling operations. (06 Marks)
  - b. Draw a neat sketch to show major parts of a horizontal milling machine. (08 Marks)
  - c. What is Indexing? Explain compound indexing. (06 Marks)
6.
  - a. List various grinding wheel abrasives and bonding processes. (07 Marks)
  - b. Sketch a centre type – cylindrical grinding machine indicating parts. (08 Marks)
  - c. Differentiate between horizontal and vertical grinding machines. (05 Marks)
7.
  - a. Sketch and explain vertical lapping machine. (08 Marks)
  - b. Name the parts along with a neat sketch of a tool head for honing of holes. (07 Marks)
  - c. List the uses of lapping process. (05 Marks)
8.
  - a. Mention any five non – traditional machining processes. (05 Marks)
  - b. Explain AJM with a neat sketch. (10 Marks)
  - c. What are the applications of non traditional machining methods? (05 Marks)

# Fitting And Machining N2 Past Question Papers

**M Mark**



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reaction a redox reaction explain your answer  $2\text{NaCl aq} + \text{Pb} \rightarrow \text{Pb} + 2\text{NaCl}$   $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$   $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + 2\text{H}_2$

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$  oxidation reduction  $25 \text{ Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent  $\text{a } \text{I}_2\text{O}_5$   
 $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$   $5 \text{ C} + \text{O}_2 \rightarrow 5 \text{ CO}_2$

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reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation  
state and oxidation number concepts to identify redox reactions and to keep track of electrons

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reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^{2+}$  oxidized reducing agent  $\text{O}_2$  to  $\text{O}^{2-}$  reduced ox ag  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$   $\text{Li}^0$  to  $\text{Li}^{+}$  oxidized red ag  
 $\text{S}^0$  to  $\text{S}^{2-}$  reduced ox ag  $3 \text{ 2Cs} + \text{Br}_2$

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