

## Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their predictable characteristic chemical behaviour. Atoms of one element cannot be transformed into atoms of another element, which demonstrates that each element's particles are made of atoms with only one type of nucleus. However, when the elements are broken down into particles, we find that they are made of atoms of different elements. We will now examine the particles that make up the atoms for which there is evidence a nucleus is made of protons and neutrons, the building blocks of the atom itself.

1. Write the formulae and draw the particle diagrams for each compound. The ratio of ions in each compound is given.

Atomic structure formulae	16 electrons 16 protons 1:1 ratio	16 electrons 16 protons 1:1 ratio	16 electrons 17 protons 1:2 ratio	16 electrons 17 protons 1:2 ratio
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particle



Diagram

Atomic structure formulae	16 electrons 16 protons 1:1 ratio	16 electrons 16 protons 1:1 ratio	16 electrons 17 protons 1:2 ratio	16 electrons 17 protons 1:2 ratio
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particle



Diagram

Atomic structure formulae	16 electrons 16 protons 1:1 ratio	16 electrons 16 protons 1:1 ratio	16 electrons 17 protons 1:2 ratio	16 electrons 17 protons 1:2 ratio
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particle



Diagram

# Modeling Chemistry Unit 6 Test Review

**Richard A. McPherson, Matthew R.  
Pincus**



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