Name \_\_\_\_\_

# Chemistry - Unit 7 Review

#### **Chemical Reaction Model**

 Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

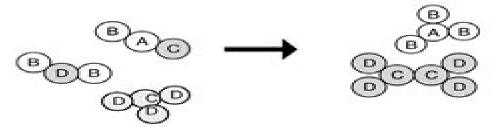
In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants

 $2 H_2 + O_2 \rightarrow 2 H_2O + energy$ 

Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

#### Applying the Model

Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_3$$

Which of these molecules would be called reactants?

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Lei Shi

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## Table of Contents Modeling Chemistry Answers Key U5 Test B V2 0

- 1. Understanding the eBook Modeling Chemistry Answers Key U5 Test B V2 0
  - The Rise of Digital Reading Modeling Chemistry Answers Key U5 Test B V2 0
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Modeling Chemistry Answers Key U5 Test B V2 0
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modeling Chemistry Answers Key U5 Test B V2 0
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Modeling Chemistry Answers Key U5 Test B V2 0
  - Personalized Recommendations

- Modeling Chemistry Answers Key U5 Test B V2 0 User Reviews and Ratings
- Modeling Chemistry Answers Key U5 Test B V2 0 and Bestseller Lists
- 5. Accessing Modeling Chemistry Answers Key U5 Test B V2 0 Free and Paid eBooks
  - Modeling Chemistry Answers Key U5 Test B V2 0 Public Domain eBooks
  - Modeling Chemistry Answers Key U5 Test B V2 0 eBook Subscription Services
  - Modeling Chemistry Answers Key U5 Test B V2 0 Budget-Friendly Options
- 6. Navigating Modeling Chemistry Answers Key U5 Test B V2 0 eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Modeling Chemistry Answers Key U5 Test B V2 0 Compatibility with Devices
  - Modeling Chemistry Answers Key U5 Test B V2 0 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modeling Chemistry Answers Key U5 Test B V2 0
  - Highlighting and Note-Taking Modeling Chemistry Answers Key U5 Test B V2 0
  - o Interactive Elements Modeling Chemistry Answers Key U5 Test B V2 0
- 8. Staying Engaged with Modeling Chemistry Answers Key U5 Test B V2 0
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modeling Chemistry Answers Key U5 Test B V2 0
- 9. Balancing eBooks and Physical Books Modeling Chemistry Answers Key U5 Test B V2 0
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modeling Chemistry Answers Key U5 Test B V2 0
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Modeling Chemistry Answers Key U5 Test B V2 0
  - Setting Reading Goals Modeling Chemistry Answers Key U5 Test B V2 0
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modeling Chemistry Answers Key U5 Test B V2 0
  - Fact-Checking eBook Content of Modeling Chemistry Answers Key U5 Test B V2 0

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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