

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Use an appropriate chemical equation, write a complete and balanced equation for each of the following reactions. Balance the equations and label the phases. The mass is underlined> (use 100% percent).

1. Chlorine gas reacts with a mixture of potassium hydroxide, KOH , and potassium chlorate, $KClO_3$, to produce potassium chloride, KCl , and a redox-neutral oxygen gas.



100% yield
 100% yield

2. Zinc metal is displaced from a solution, copper(II) sulfate pentahydrate solution, produce copper(II) sulfate and zinc ions. How many moles of copper(II) sulfate will be produced by reacting 25.00 g of zinc metal?



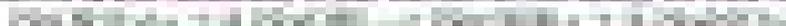
100% yield
 100% yield

3. Sulfuric, H_2SO_4 , reacts with copper(II) carbonate to produce sulfur dioxide gas and copper(II) sulfate. How many moles of copper(II) sulfate are produced by reacting 10.00 g of copper(II) carbonate?



100% yield
 100% yield

4. Chlorine gas reacts with a mixture of sodium hydroxide and sodium chlorate to produce sodium chloride, sodium sulfate, and sodium chlorate.



100% yield
 100% yield

Modeling Chemistry Unit 8 4

Padhraic Smyth



Modeling Chemistry Unit 8 4:

Unveiling the Magic of Words: A Overview of "**Modeling Chemistry Unit 8 4**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Modeling Chemistry Unit 8 4**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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