

Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

- a. Aluminum metal reacts with hydrochloric acid to form aluminum chloride.
- b. Acetylene (C_2H_2), a hydrocarbon, burns with oxygen (O_2) and carbon dioxide.
- c. Zinc reacts with copper(II) sulfate to form zinc sulfate and copper.
- d. Ethanol reacts with oxygen to form carbon dioxide and water.
- e. Sodium metal reacts with water to form sodium hydroxide.
- f. Nitrogen (N_2) is placed in a hot glowing crucible and reacts with the gas.
- g. Magnesium metal burns when exposed to light to form magnesium hydroxide.
- h. Nitrogen gas reacts with some red glowing metal to form Mg_3N_2 .

Modeling Chemistry Unit 7 Ws3 Answers

Staley Wilbraham (Matta, Waterman)



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