

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their consistently different mass. However, when the elements are broken apart by their respective chemical reactions, the atoms are not broken apart, but instead combine to form compounds. Therefore, when the elements are broken apart by their respective chemical reactions, the atoms are not broken apart, but instead combine to form compounds. We will now examine the patterns that exist for the atoms in which these elements combine to form the compounds that are the basis of life.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic symbols and formulas	12 hydrogen 16 oxygen H_2O	12 hydrogen 16 oxygen H_2O_2	12 hydrogen 1 carbon CH_4	1 hydrogen 1 carbon CH
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particle



diagram

Atomic symbols and formulas	12 carbon 16 oxygen C_2O	12 hydrogen 16 oxygen H_2O_2	12 carbon 1 oxygen C_2O
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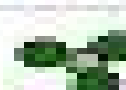
particle



diagram

Atomic symbols and formulas	12 hydrogen 16 carbon H_2C_2	12 hydrogen 16 carbon H_2C_2	1 hydrogen 16 carbon HC_2	1 hydrogen 1 carbon HC
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particle



diagram

Modeling Chemistry Unit 6 Test Answer Key

E Durkheim



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