

Chapter 16: Equilibrium-Related problems

Section 16.1 - 16.2

1. Write the equilibrium constant expression for K_c for the following reactions:



2. Write the equilibrium constant expression for K_p for $\text{H}_2\text{(g)} + \text{Cl}_2\text{(g)} \rightleftharpoons 2\text{HCl(g)}$

3. Write the chemical equation for ammonia formation and hydrogen, K_c is 0.50 at 500°C.



Calculate K_p for this reaction at the temperature

4. If at equilibrium $2\text{NO}_2(\text{g}) \rightleftharpoons \text{N}_2\text{O}_4(\text{g}) + \text{O}_2(\text{g})$ at a temperature of 2500 K, the mass of oxygen at 0.050 mol O_2 . Calculate the value of K_p .

5. Write the equilibrium constant expressions for the acid-base pairs of the following reactions:



Equilibrium Ap Test Practice Problems Answers

Neil D. Jespersen, Pamela Kerrigan



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