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ANSWERKEY	To all the second	out of 39 points.

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Dute	Period

# Electron Dot Diagram Periodic Table

Use a periodic table to construct Electron Dot Diagrams (Lewis Structure) for elements with atomic numbers 1-18.

Each box is worth one point.

H							i • • He
Li	°•• Ве	B	° C	· N	:0	°F•	•Ne
"• Na	Mg	٩İ	Si	• P	: S	:CI·	:Ar:

#### Follow-up Questions:

List your observations about the periodic table above: (4 points)

Suggested Answers: There are patterns. Elements in the same column (group family) have the same number of dots (valence electrons). Elements in Group 1 have one valence electron; elements in Group 2 have two valence electrons; elements in Group 13 have three valence electrons; elements in Group 14 have four valence electrons, etc. As you move across the period from left to right, one dot (electron in the last energy level) is added.

# **Electrons And The Periodic Tabke Andwer Keh**

**Emilie Sanchez** 

#### **Electrons And The Periodic Tabke Andwer Keh:**

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