

# Periodic Table Groups and Periods

A group is a column on the periodic table. Elements have the same number of valence electrons.

A period is a row on the periodic table. Elements have the same number of electron shells.

There are 7 periods.

There are 18 groups.

The periodic table is color-coded by groups: Group 1 (blue), Group 2 (yellow), Groups 3-10 (yellow), Group 11 (yellow), Group 12 (yellow), Groups 13-18 (teal), and Groups 19-20 (pink). The lanthanide and actinide series are shown below the main table.

1	2																	18																	
1	H																	2	He																
3	Li	4	Be																	10	Ne														
5	Na	6	Mg																	16	Ar														
7	K	8	Ca	9	Sc	11	Ti	12	V	13	Cr	14	Mn	15	Fe	16	Co	17	Ni	18	Cu	19	Zn	20	Ga	21	Ge	22	As	23	Se	24	Br	25	Kr
9	Rb	10	Sr	11	Y	13	Zr	14	Nb	15	Mo	16	Tc	17	Ru	18	Rh	19	Pd	20	Ag	21	Cd	22	In	23	Sn	24	Sb	25	Te	26	I	27	Xe
11	Cs	12	Ba	13	La	14	Hf	15	Ta	16	W	17	Re	18	Os	19	Ir	20	Pt	21	Au	22	Hg	23	Tl	24	Pb	25	Bi	26	Po	27	At	28	Rn
13	Fr	14	Ra	15	Ac	16	Rf	17	Db	18	Sg	19	Bh	20	Hs	21	Mt	22	Ds	23	Rg	24	Cn	25	Nh	26	Fl	27	Mc	28	Lv	29	Ts	30	Og
57	La	58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb	71	Lu						
89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Md	102	No	103	Lr						

# Electrons And Periodicity Packet Answers

**M Mark**



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