

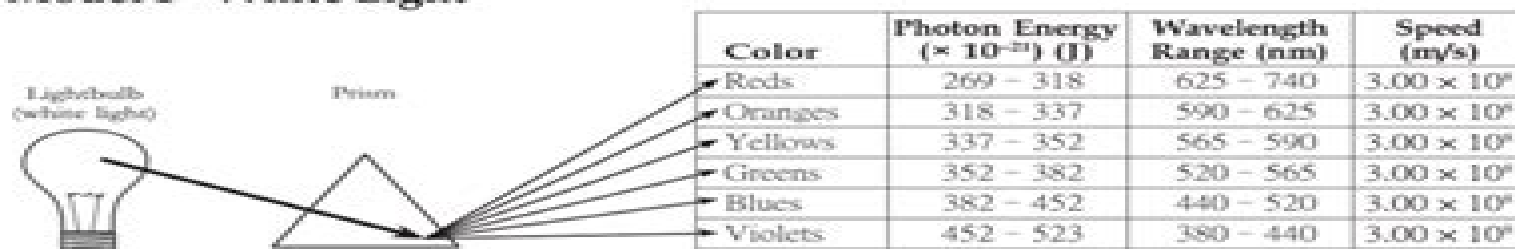
Electron Energy and Light (Electron Excitation)

Why?

How does light reveal the behavior of electrons in an atom?

From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms. Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

Model 1 – White Light



- Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- What happens to white light when it passes through a prism?
- Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?
- Do all colors of light travel at the same speed?
- Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?
- Consider the light illustrated in Model 1.
 - Which color corresponds to the longest wavelengths?
 - Which color corresponds to the shortest wavelengths?
 - Write a sentence that describes the relationship between wavelength and energy of light.



Electron Energy And Light Worksheet Answers

Stewart C. Bushong



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Radiologic Science for Technologists Stewart C. Bushong, 1997 The purpose of this textbook is to convey a working knowledge of radiologic physics and to prepare radiography students for the certification exam by the ARRT The textbook also provides a standard of knowledge from which practicing radiographers can make decisions about technical factors and diagnostic image quality in the work place This edition gives an expanded coverage of quality management which includes all of the content on the ARRT It also includes coverage of new cardiovascular interventional equipment and recent advances in spiral CT and digital radiography Keeps students informed and up to date with respect to professional standards and requirements

The Nature of Matter Gr. 5-8 , Radiologic Science Stewart C. Bushong, 1984 **Project SPICA** Nadine Butcher Ball, Harold P. Coyle, Irwin I. Shapiro, 1994 **Spotlight Science** Keith Johnson, Sue Adamson, Gareth Williams, 2002 This Spiral Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department

Journal of Biological Education , 1985 *Directory of Harvard Business School Cases and Related Course Materials* Harvard University. Graduate School of Business Administration, 1985 *Subject Guide to Books in Print* , 1975 *Electron Energy Levels* , 1962

Electron Energy Levels in Ordered and Disordered Structures Marilyn E. Usher, 1994 **Energy Loss and Range of Electrons and Positrons** Ann T. Nelms, 1956 *Electron energy levels in crystalline solids* David K. Adams, 1952

Electron Energy and Position Measurement Using Lead-scintillator Calorimeters with a New Light Collection System W. Hoffmann, D. Wegener, J. Spengler, 1978 **Electron Energy in Solids** Joseph Callaway, 1958 Electron Energy and Mean Forward Slowing Down Approximation , 1979 **Electron Energy Loss Spectra of Atoms and Molecules** , 1970

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