

Unit 5 – Electrochemistry Test

NAME: Emanuel Suissa

K/U	C	A	T/I
/8	/10	/7	/4

Part A: Knowledge & Understanding – Multiple Choice (8 Marks)

Identify the choice that best completes the statement or answers the question

- Which of the following substances would make a poor electrolyte solution?
a) Lead (II) Nitrate ☒ c) Silver Bromide
✓ b) Potassium Chloride d) Ammonium Phosphate
- Which of the following solid metals could act as the anode if paired with cadmium ($\text{Cd(s)}/\text{Cd}^{2+}$) in a galvanic cell?
a) Lead ☒ c) Zinc
b) Copper d) Tin
- The proper cell notation for a galvanic cell composed of a chromium electrode and a cadmium electrode would be:
→ ☒ a) $\text{Cr(s)} | \text{Cr}^{3+}(\text{aq}) || \text{Cd}^{2+}(\text{aq}) | \text{Cd(s)}$
b) $\text{Cd(s)} | \text{Cd}^{2+}(\text{aq}) || \text{Cr}^{3+}(\text{aq}) | \text{Cr(s)}$
c) $\text{Cr}^{3+}(\text{aq}) | \text{Cr(s)} || \text{Cd(s)} | \text{Cd}^{2+}(\text{aq})$
d) $\text{Cd}^{2+}(\text{aq}) | \text{Cd(s)} || \text{Cr(s)} | \text{Cr}^{3+}(\text{aq})$
- Which of the following metals is the least susceptible to rusting?
a) Nickel
b) Copper
✓ ☒ c) Gold
d) Zinc
- Which of the following is an example of an alloy?
a) Bromine
b) Iron
→ ☒ c) Bronze
d) Copper Nitrate

Electrochemistry Test A Answer

LP Steffe

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'LP Steffe' text.

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