Chemistry Unit 3 Test Review

5.1 Mole and Avogadro's Constant

A particle is very small and can be:

- A single atom (Ca)
- A molecule with 2 or more atoms (molecular compound H2O)
- An ion with a positive or negative charge (CO₁2-)
- A formula unit with two or more ions (ionic compound NaCl)

The Mole

- · Chemists use the mole to represent a specific number of particles
- The number of particles it represents is called the Avogadro constant (6.02 × 10²⁹ particles/mol)
- The unit for mole is mol

Definitions

- Mole: One mole of any entity is 6.02 × 10²⁵ of the particles (atom, molecule, ion formula unit); the symbol is n; the unit is mol
- Avogadro's constant: The number of particles in one mole, 6.02 × 10²⁸; the symbol is Na; the unit is particles/mol
- 3. Particles:
- Atom or element (Eg. Zn)
- Molecule or covalent compound (Eg. H2O)
- lons or monatomic/polyatomic ions (Eg. Zn+2 and CO₁2-)
- Formula unit or ionic compound (Eg. LiCl)

5.2 Molar Mass

Basics: The Mass and The Mole

- The mass of carbon-12 is 12 atomic mass unit (12 u). Every other element has mass relative to one atom of carbon-12
- 1 mole of carbon-12 = 6.02 × 10²³ atoms of C-12 = 12g
- The average atomic mass of any element, in atomic mass units, is numerically
 equivalent to the mass of one mole of atoms (6.02 × 10²⁸ atoms) of that element

Motar Mass: The mass of one mole of its particle – atoms, molecules, ions, or formula units, in grams per mole (g/mol)

6.1 Percentage Composition

Law of Definite Proportions:

- · Elements in a chemical compound are always present in the same proportions by mass
- Law of constant composition: all molecules or formula units of a compound will have the same chemical composition

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U.S. Nuclear Regulatory Commission

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U.S. Nuclear Waste Technical Review Board Report to the U.S. Congress and the Secretary of Energy: January to December 1998, Advanced Computational Approaches for Drying in Food Processing Krunal M. Gangawane, Madhuresh Dwivedi, Ram Chandra Pradhan, 2024-09-17 Computational methods have become important techniques for drying in food processing There are two principle computational approaches for system analysis continuous and discrete In the continuous approach the governing equations can be obtained by applying the fundamental laws such as conservation of mass momentum and energy over an infinitesimal control volume These equations are further discretized by using a suitable discretization technique The recovered set of algebraic equations are then solved by an applied numerical method The discrete approach concentrates on mimicking the molecular movement within system Recent years have witnessed a rapid development in the field of computational techniques owing to its abundant benefit to the food processing industry The relevance of advanced computational methods has helped in understanding the fundamental physics of thermal and hydrodynamics behavior that can provide benefits to the food processing industry in numerous applications such as drying evaporation sterilization mixing and refrigeration Advanced Computational Approaches for Drying in Food Processing examines the use of different numerical computational techniques for the simulation of fluid flow and heat and mass transfer from within food products such as cereal chicken beef fruits vegetables and more The text promotes a thorough

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