

2. Three dash lines (_____) → _____
 = _____ / _____ make up bonds (_____)

Example:



III. Properties of Organic Compounds

- A. Bonding: _____ → _____ = _____
 B. Solubility: most are _____ in water (generally _____)
 = _____
 C. Conductivity: mostly _____ (s), (l), & (aq) states
 *Only _____ in solution = _____
 D. Melting/boiling points: _____ → _____
 E. Reactivity Rate: _____; covalent molecules tend to
 have relatively _____ → _____ in
 reaction → not taking longer

IV. Types Of Chemical Formulas

- A. Molecular Formula: shows the _____ of each _____
 in a compound; least informative formula

Ex: Propane = C_3H_8

- B. Structural Formula: shows the _____ of each _____
AND the _____ of the _____; most informative formula



- C. Condensed Formula = _____ of both _____ and
 _____ formulas: each carbon is written with its constituent
 hydrogens followed by the proper subscript

Ex: Propane = $\text{CH}_3\text{CH}_2\text{CH}_3$

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