

Chemistry – Unit 5 Worksheet 1

Relative Mass

Relative Mass from Gases

We have established that the combining ratio of gases can be explained if two assumptions are made:

1. Equal volumes of gases contain the same number of molecules at the same pressure and temperature.
2. Some pure elemental gases are clustered into pairs to form diatomic molecules.

Use these assumptions and particle diagrams to explain the fact that the density of oxygen gas at standard temperature and pressure is 1.43 g/L, whereas the density of hydrogen gas under these conditions is 0.089g/L. How many times more massive is one molecule of oxygen than one molecule of hydrogen?



You shouldn't conclude that chemists were able to determine the molar masses of all elements using this technique. Measurements of the density of the gaseous phase of many of the elements are difficult, if not impossible. However, we are going to see that chemists use another tool – the *percent composition of compounds* to determine molar masses.

Relative Mass from Compounds

Many substances combine with oxygen to form a type of compound called an oxide. We have already seen that such combinations often occur in multiple proportions. John Dalton made the assumption that the lowest mass ratio that could be found for a compound was achieved when the elements combined in a 1:1 combination. For now, we will make a similar assumption.

In Unit 4 you examined the percent composition data of several compounds. For example, 100 g of the simplest oxide of carbon contains 42.9 g of carbon and 57.5 g of oxygen. It is helpful when comparing masses of elements, to compare them to the same mass of oxygen. Using a proportion, one can show that 75.0 g of carbon combines with 100 g of oxygen.

$$\frac{42.9\text{g C}}{57.1\text{g O}} = \frac{75.0\text{g C}}{100\text{g O}}$$

Modeling Chemistry U6 Ws 5 Vanswers

**Douglas Buchanan, James Melrose, A. R.
Ross**



Modeling Chemistry U6 Ws 5 Vanswers:

Answers to Revision Questions for National 5 Chemistry Douglas Buchanan, James Melrose, A. R. Ross, 2013

The Top Books of the Year Modeling Chemistry U6 Ws 5 Vanswers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the engaging narratives that have enthralled audiences this year. Modeling Chemistry U6 Ws 5 Vanswers : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Modeling Chemistry U6 Ws 5 Vanswers : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Popen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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