

Chemistry – Unit 6 Worksheet 1

After having observed molecular models which show 18.098 compounds we discussed the structural formulae based on formal positively charged ions, local formal negative formal charges, formal for formal negatively charged ions (anions). However, many chemical species exhibit structural formulae with a different set of formal charges which are not necessarily effects to be ignored. After having these examples the students must avoid that they make formal charges elements according to the the differences the charges of the ions they have.

1. After that formulae and draw the particular structure that shows a compound. The name and ions are also represented in given.

18.098 structural formulae	1. structure 2. oxygen	2. structure 3. oxygen	3. structure 4. oxygen	4. structure 5. oxygen
	H_2O	H_2O	H_2O	H_2O

particular
structure



18.098 structural formulae	2. structure 3. oxygen	3. structure 4. oxygen	4. structure 5. oxygen
	H_2O	H_2O	H_2O

particular
structure



18.098 structural formulae	2. structure 3. oxygen	3. structure 4. oxygen	4. structure 5. oxygen	5. structure 6. oxygen
	H_2O	H_2O	H_2O	H_2O

particular
structure



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L Cohen



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