

Name: _____

Period: _____

Mixed Mole Problems Worksheet

Directions: Solve the following problems. Be sure to show all your work.

1. Find the number of moles if there are 7.63×10^{24} molecules of a substance.
2. How many moles of CaCO_3 are there in 2.00 kg of CaCO_3 ?
3. What is the mass of 8.42 moles of $(\text{NH}_4)_2\text{S}$?
4. Find the number of carbon atoms in 18.5 g of CH_3OH (methanol).
5. If one litre of a gas at STP has a mass of 3.17 g, what is the molar mass of this gas?
Name the diatomic element that makes up this gas?
6. What volume at STP would 11.50 g of oxygen gas occupy?
7. What is the mass of 33.2 L of SO_2 gas at STP?
8. Calculate the molar mass of a gas at STP, if 360.0 mL of this gas has a mass of 0.680 g

Mixed Mole Problems Show Work

Wenbin Ji



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