



Ford Ranger 2007 Manual

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assuming CeCl_2 CuCl_2 CuCl_2 is available in excess

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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr}$
 using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$
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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr}$
 SnF_2

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 $\text{SiO}_2 + \text{C} \rightarrow \text{SiC} + \text{CO}$

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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$
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 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr}$
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