



JANE ASHFORD

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P.-F. Puisieux



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, **Vie et mort du couple en Nouvelle-France** Josette Brun, 2006-11-15 La Nouvelle France offre t elle aux femmes un champ d action largi comme le voudrait une certaine conception de l histoire coloniale Ce n est pas ce que r v le l analyse du partage des droits et des responsabilit s entre poux des secondes noces et des strat gies de survie conomique des personnes veuves Ma tres et seigneurs chez eux selon le v u de l tat de l glise et de la loi les maris assument formellement l essentiel des responsabilit s professionnelles et patrimoniales du m nage Lorsqu ils meurent leurs veuves doivent pour survivre apprendre profiter de leur nouvelle capacit juridique d une certaine flexibilit des r les f minins de leur exp rience professionnelle ou de leur douaire qu elles mettent en valeur seules ou avec l aide de la parent Les veufs souvent parents de jeunes orphelins et contraints par les normes de la masculinit se remarient rapidement Ils retiennent moins l attention des autorit s que les veuves nombreuses dom la figure tant t attendrissante et tant t suspecte se trouve parfois directement m l e aux rapports de pouvoir entre la m tropole et ses colonies nord am ricaines

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web elements in the second row of the periodic table place their electrons in the $2n$ shell as well as the $1n$ shell for instance lithium li has three electrons two fill the $1s$ orbital and the third is placed in the $2s$ orbital giving an electron configuration of $1s^2 2s^1$

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web electron configurations and trends answer these discussion questions given the following part of an electron configuration $3s^2$ explain which part refers to the number of electrons the energy level and the sublevel what is the maximum number of electrons that can occupy a single orbital state and explain hund's rule

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he $2s^2$ is an s electron whereas the electron removed during the ionization of boron he $2s^2 2p^1$ is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 1 or simply put as 4s 1 q2 what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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web electron configurations are a simple way of writing down the locations of all of the electrons in an atom as we know the positively charged protons in the nucleus of an atom tend to attract negatively charged electrons

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