

Chapter 1

Introduction to Electronics

Question A.1. Atomic Structure

- (a) An atom is the smallest particle of an element that cannot be created or destroyed.
- (b) The atom consists of three subparticles, i.e., electron, proton and neutron.

Question A.2. Materials Used for Electronics

- (a) The materials are generally classified into three categories: (i) conductors, (ii) semiconductors, and (iii) insulators.
- (b) The atoms containing valence electrons are called conductors.
- (c) An atom having four or five valence electrons is called a semiconductor.

Question A.3. Current in Semiconductors

- (a) Movement of electrons from one electron to the next is called drift current.
- (b) Current is produced when there is a potential difference across the material.

Question A.4. N-Type and P-Type Semiconductors

- (a) In an n-type semiconductor, the atoms of the semiconductor are replaced by pentavalent atoms to produce negative charge carriers (electrons) in the material.
- (b) In a p-type semiconductor, the atoms of the semiconductor are replaced by trivalent atoms to produce positive charge carriers (holes) in the material.

Question A.5. The PN Junction

- (a) The junction is formed by joining two different semiconductor materials. The junction is formed by joining an n-type semiconductor material with a p-type semiconductor material. The junction is formed by joining an n-type semiconductor material with a p-type semiconductor material.
- (b) The junction is formed by joining two different semiconductor materials. The junction is formed by joining an n-type semiconductor material with a p-type semiconductor material.

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