

Name KEY

Orbital Diagrams

Orbital diagrams show the electronic configuration and electron pairings within an atom. Each box represents an orbital. Each orbital can only hold two electrons. Electrons are represented by arrows, which can point up (spin up) or point down (spin down). Remember the two rules for orbital diagrams: *Pauli Exclusion Principle* - no two electrons with the same spin can occupy the same orbital, and *Hund's Rule* - every orbital in the same sublevel must have one electron before electrons will pair.

First label the boxes with their principal energy level number and sublevel letter. Then start to fill in the electrons (put arrows in the boxes). When your orbital diagram is finished, write the final electron configuration on the line below the orbital diagram.



Draw an orbital diagram for the following atoms on your own.



Electron Configuration Lab Answer Key

G Psacharopoulos



Electron Configuration Lab Answer Key:

Electron Configuration Lab Answer Key Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Electron Configuration Lab Answer Key**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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