

8 • Electron Configurations & Periodicity

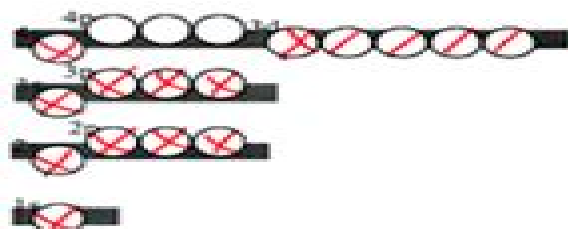
WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
Element: Ar # of e ⁻ s: <u>18</u>	Element: Mg # of e ⁻ s: <u>12</u>	Element: N # of e ⁻ s: <u>7</u>	Element: Li # of e ⁻ s: <u>3</u>	Element: P # of e ⁻ s: <u>15</u>	Element: Cl # of e ⁻ s: <u>17</u>

Write the electron configurations of each of these in **long form** and **short form**:

- Ar $1s^2 2s^2 2p^6 3s^2 3p^6$
 Ar $[Ne] 3s^2 3p^6$
- Mg $1s^2 2s^2 2p^6 3s^2$
 Mg $[Ne] 3s^2$
- N $1s^2 2s^2 2p^3$
 N $[He] 2s^2 2p^3$
- Li $1s^2 2s^1$
 Li $[He] 2s^1$
- P $1s^2 2s^2 2p^6 3s^2 3p^3$
 P $[Ne] 3s^2 3p^3$
- Cl $1s^2 2s^2 2p^6 3s^2 3p^5$
 Cl $[Ne] 3s^2 3p^5$



- Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.



Electron Arrangements In Atoms Worksheet Answers

DP Hallahan



Electron Arrangements In Atoms Worksheet Answers:

Whispering the Techniques of Language: An Mental Quest through **Electron Arrangements In Atoms Worksheet Answers**

In a digitally-driven world wherever screens reign great and instant conversation drowns out the subtleties of language, the profound strategies and mental subtleties concealed within phrases usually go unheard. Yet, set within the pages of **Electron Arrangements In Atoms Worksheet Answers** a fascinating literary value sporting with natural emotions, lies a fantastic journey waiting to be undertaken. Composed by an experienced wordsmith, that marvelous opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within ab muscles fabric of each and every word. Within the mental depths of the touching evaluation, we will embark upon a genuine exploration of the book is key subjects, dissect its interesting writing style, and yield to the powerful resonance it evokes heavy within the recesses of readers hearts.

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