

CHAPTER 4 REVIEW*Arrangement of Electrons in Atoms***MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

- 1.**
- Under what conditions is a photon emitted from an atom?

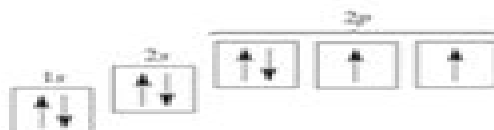
A photon is emitted when an electron moves from a higher energy level to a lower energy level.

- 2.**
- What do quantum numbers describe?

Quantum numbers describe the energy level, orbital shape, orbital orientation, and spin state of an electron.

- 3.**
- What is the relationship between the principal quantum number and the electron configuration?

The principle quantum number, n , describes the energy level. For example, the electrons at $2p^6$ are at the energy level represented by $n = 2$.



- 4.**
- In what way does the figure above illustrate Hund's rule?

The most stable arrangement of electrons is one with the maximum number of unpaired electrons.

- 5.**
- In what way does the figure above illustrate the Pauli exclusion principle?

No two electrons have the same set of four quantum numbers.

Electron Arrangement Review Worksheet Answers

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