

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Knows through that only being composed exclusively of the carbon and hydrogen atoms in their state
- Natural abundant (petroleum, natural gas, etc.) or synthetic hydrocarbons are synthesized, present organic compounds in nature
- Today, organic chemistry includes the discovery of almost all carbon compounds, regardless of origin

Introduction to Hydrocarbons

Hydrocarbon compounds are called hydrocarbons

Hydrocarbon is an organic compound consisting of carbon and hydrogen

Hydrocarbon from their molecular structure is called when always from carbon bonds

Hydrocarbon hydrocarbons are saturated alkanes, alkenes and alkynes, which are called when they contain all carbon bonds

Alkanes is the simplest and also most important of organic compounds. Alkanes have only a single carbon-carbon bond in their molecules

→ They always are formed by four carbon bonds, which means, in this sense that carbon is the only sufficient compound



The properties of hydrocarbons

Hydrocarbons are organic molecules

Hydrocarbons from the chemical structure that they have, i.e., the hydrocarbons with low molecular weight are the gas phase, hydrocarbons with high molecular weight

How do hydrocarbons differ?

Hydrocarbons are divided into alkanes, alkenes, alkynes, etc. and their properties

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