

# Hydrocarbons

Methane	General formula for an alkane	General formula for an alkene	Ethene	Propene	Butene
Ethane	How to test between an alkene and alkane	Alkane or alkene?  C <sub>2</sub> H <sub>6</sub> C <sub>3</sub> H <sub>8</sub> C <sub>4</sub> H <sub>10</sub> C <sub>5</sub> H <sub>12</sub>	Show how ethene is polymerised		
Propane	Definition of a hydrocarbon		What is meant by saturated/unsaturated		
Butane	How ethanol is produced... ...from ethane     ...by fermentation		Draw the monomer that makes this polymer $  \begin{array}{cccccc}  \text{Br} & \text{H} & \text{Br} & \text{H} & \text{Br} & \text{H} \\    &   &   &   &   &   \\  \text{H}-\text{C}-\text{C}-\text{C}-\text{C}-\text{C}-\text{C}-\text{H} \\    &   &   &   &   &   \\  \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H}  \end{array}  $		
A way to remember the order of alkanes			Describe the process of cracking Process:   Types of products produced:  Conditions:  Why it is done:		Complete the equations for cracking $\text{C}_{10}\text{H}_{22} \rightarrow \text{C}_2\text{H}_6 + \text{_____} + \text{C}_6\text{H}_{14}$ $\text{C}_8\text{H}_{18} \rightarrow \text{C}_3\text{H}_8 + \text{_____}$ The types of products produced:

# Hydrocarbon Compounds Core Teaching Resources

**Bornok Sinaga, Darwin, Juniastel  
Rajagukguk**



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