

**Chapter**  
**Thermodynamics: Kinetic Molecular**  
**Theory and Calorimetry**

••• The **state** of a system is described by **macroscopic** **variables** such as **temperature** and **volume**.

••• **State** variables are **extensive** variables, unlike the **intensive** of the **state**. **State** **variables** are **intensive**.

**4. Kinetic Molecular Theory**

1. **Define** the **kinetic theory**.

**The study of particles and their motion.**

2. **List** the **fundamental** **assumptions**.

**1. Matter is made up of particles.**

2. **Particles** are **in constant** **motion** and **collide** with **each other**.

**3. Particles** are **in constant** **motion** and **collide** with **each other** and **the walls** of the **container**.

**4. Particles** are **in constant** **motion** and **collide** with **each other** and **the walls** of the **container**.

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3. **List** the **fundamental** **assumptions** of **kinetic theory**.

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