Stoichiometry - Gram Conversions

Use the following balanced equation to answer the questions below:

- a. How many grams of O₁ are produced when 34.5g of HgO decompose?
- b. How many grams of HgC need to decompose to produce 0.277g of Hg?
- c. If 82 tg of HigO decomposes, then how many grams of Hig will be made?
- When catalyzed by sulfuric acid, potassium chlorate and sucrose react to produce heat, CO_b, M₂O_c and RCI. Below is a balanced equation for this reaction.

- a. How many grams of CO, will form if 52.9g of sucrose react completely with KCIO,?
- How many grams of KCIO, are needed to completely react with 781g of C.,H.,O.,7
- c. How many motes of water will be produced if 99.1g of KCIO₆ reacts completely with suprose?
- Nitrogen dicicide reacts with water to produce mitric acid and ratric civide. Below is the baterioed equation for this reaction.

a. How many grams of NO will be made if 8.67g of NO, reacts completely with water?

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Ying-Ying Zheng

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