

The label shows the ingredients in a drink called Cola.

Cola
Ingredients:
Carbonated water
Sugar
Colouring
Phosphoric acid
Flavouring
Caffeine

1 (a) (i) The pH of carbonated water is 4.5.

The pH of Cola is 2.9.

Name the ingredient on the label that lowers the pH of Cola to 2.9.

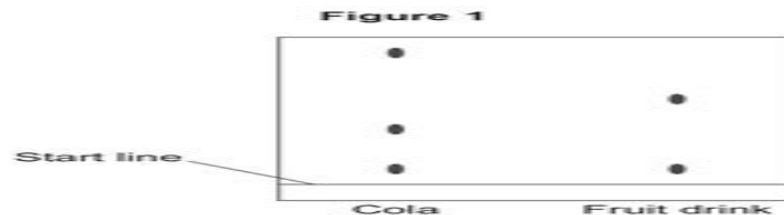
[1 mark]

1 (a) (ii) Which ion causes the pH to be 2.9?

[1 mark]

1 (b) A student investigated the food colouring in Cola and in a fruit drink using paper chromatography.

The chromatogram in Figure 1 shows the student's results.



1 (b) (i) Complete the sentence.

The start line should be drawn with a ruler and

Give a reason for your answer.

[2 marks]

1 (b) (ii) Suggest **three** conclusions you can make from the student's results.

[3 marks]

1 (c) Caffeine can be separated from the other compounds in the drink by gas chromatography.

Why do different compounds separate in a gas chromatography column?

[1 mark]

1 (d) Caffeine is a stimulant.

Large amounts of caffeine can be harmful.

1 (d) (i) Only **one** of the questions in the table **can** be answered by science alone.

Tick (✓) **one** question.

[1 mark]

Question	Tick (✓)
Should caffeine be an ingredient in drinks?	
Is there caffeine in a certain brand of drink?	
How much caffeine should people drink?	

1 (d) (ii) Give **two** reasons why the other questions **cannot** be answered by science alone.

[2 marks]

Grademathematics Novembes 2014 Paper

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