

ঘের ও কীটপতঙ্গের অস্বাস্থ্যের নির্বাহী পদ্ধতির প্রয়োজন

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে উল্লিখিত X-পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ ঘন ও Y-পদ্ধতিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

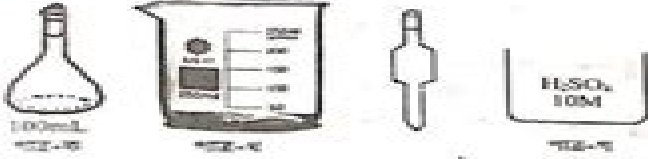
যদিও X-পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদ্ধতিটি হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 মিলিগ্রাম রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে কাজ করা হয়। ফলে পরিবেশে মৃদু উপযোগযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকমিলিগ্রাম ব্যবহার করা হয় ফলে পরিবেশে মৃদু ঘটে না।

$(\text{CN})_2 + \text{H}_2\text{O} \rightarrow \text{HCN} + \text{HCN}$
 ফলে কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

উদাহরণ



১. উদাহরণে দু'ক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাণ্ডলার প্রক্রিয়াটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অস্বাস্থ্যের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ ও নং প্রশ্নের উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি স্কেলের জন্য যে ছবি পাওয়া যায়, তাকে রাইডার ছবি বলা হয়।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 মিলিগ্রাম রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধ বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে কাজ করা হয়। ফলে পরিবেশে মৃদু উপযোগযোগ্যভাবে কমে আসে।

অতএব, কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M দ্রবণ। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M দ্রবণ হলে পাত্র H_2SO_4 দ্রবণ। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাণ্ডলার প্রক্রিয়ায় অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাণ্ডলার প্রক্রিয়ায় অঙ্কন করে ব্যাখ্যা করা হলো:

প্রক্রিয়া	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান কলতে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে খুব কম পরিমাণে তরল ও অদ্রবণীয় পদার্থের মিশ্রণে স্থাপন করা হয়। এতে তরল পাত্রের জন্য এবং অদ্রবণীয় পদার্থের জন্য পাত্র।
উপাদান	পাত্র বীমের এলিমেন্টের (HCl , HNO_3 , H_2SO_4) পাত্র ক্ষয় (NaOH , KOH , H_2O_2) জারকীয় পদার্থের দ্রবণ ইত্যাদি।
সংরক্ষণ পাত্র	- পাত্র এলিমেন্টের পাত্রের মধ্যে রাখা হয়। - পাত্র বীমের মধ্যে রাখা হয়।

উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে হবে।

1ম চলকের দ্রবণটি হলো 100 ml জারকীয় দ্রবণ ও এর দ্রবণ পাত্র। সেমি-মাইক্রো জারকীয় দ্রবণের আয়তন 100 ml। তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে। দ্রবণের মৃদুপের দ্রবণ হতে আমরা জানি,

$$V_1 S_1 = V_2 S_2$$

$$\text{যা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ ml}$$

$$= 10 \text{ ml}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পাত্র বীম রাখতে হবে (100-10) ml = 90 ml।

আবার, পাত্রের মধ্যে 10 ml দ্রবণ ঘনমাত্রার দ্রবণ নিয়ে তার সাথে জারকীয় দ্রবণ 90 ml পাত্র বীম রাখলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার দ্রবণ পাওয়া যাবে।

উদাহরণ



১. অ্যাসিডিকিটি অঙ্কন করে।
২. সোডা ছবি K, এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যাণ্ডলার উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক দ্রবণের ব্যবহারকালে খাওয়া ঝুঁকি হলে কবীর উপর সম্পর্কিত বিশ্লেষণ করে। ৪ ও নং প্রশ্নের উত্তর

অ্যাসিডিকিটি এক ধরনের রাসায়নিক বা কৃত্রিম দ্রবণ সঞ্চেপ বা মৃদুপ বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জারন থেকে রক্ষা করে রাখার সঞ্চেপ করে।

সামান্য K, এর ঘন কখনো অসীম বা শূন্য হতে পারে না। তাইটি ব্যাখ্যা করার জন্য নিম্নের উত্তরদ্রব্য মাইক্রোপদ্ধতি বিশ্লেষণ করি-

$$A + B \rightarrow C + D$$

Hsc Chemistry 1 First Chapter Notes

JE Gale



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