

Worksheet # 1 & Gram Formula Mass Answers

- | | |
|--|------------------|
| 1. 4.5×10^{23} mcls H_2O | 1. 158.04 g/mol |
| 2. 3×10^{24} atoms Fe | 2. 74.55 g/mol |
| 3. 5.73×10^{-4} mol Cs | 3. 142.05 g/mol |
| 4. 1.1×10^5 mol Na | 4. 164.10 g/mol |
| 5. 2.1×10^{23} mcls $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ | 5. 342.17 g/mol |
| 6. 9.55×10^{-2} mol Cu | 6. 149.12 g/mol |
| 7. 1.204×10^{24} atoms Fr | 7. 249.70 g/mol |
| 8. 2.000 mol He | 8. 262.87 g/mol |
| 9. 5.6×10^2 mol S | 9. 219.52 g/mol |
| | 10. 458.16 g/mol |

Gram Formula Mass Answers

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Mascetta, 1998 An extensive review of chemistry topics a full length diagnostic test precedes four up to date full length model SAT II Chemistry Tests All tests are followed with answers charts students can use to diagnose their individual study needs Chemistry topics reviewed include atomic structure bonding all other topics included in the SAT II test Joseph A Mascetta is a science educational consultant a former principal chemistry teacher at Mount Lebanon High School Pittsburgh PA

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