

Objective: Students will identify and differentiate between the different parts and models of the atom.

1. Atoms are made of three basic particles: _____
2. What is electric charge? _____
3. A positive and a negative charge will _____
4. Two positive charges will _____
5. Two negative charges will _____
6. What does the term *ion* represent? _____
7. Electric charge only appears in _____ of the elementary charge.
8. What did J.J. Thompson discover? _____
9. Atoms have _____ charge.
10. The model thought electrons were sprinkled around inside atoms like _____ in a _____ of _____.
11. This was the _____ ball model for the _____ of an _____, but it wasn't the _____ model.
12. In 1911 Rutherford, Geiger and Marsden did an experiment where they shot helium at a thin gold foil. They expected most of the helium to be deflected a little. What actually happened? _____

13. The best way to explain this was if a gold atom was mostly _____.
14. How much of an atom's mass is contained in the nucleus? _____
15. Where is the electron found? _____
16. Where is the proton found? _____
17. Where is the neutron found? _____
18. How much bigger is the electron cloud than the nucleus? _____
19. If the atom were a football stadium, how big would the nucleus be? _____
How big would the electron be? _____
20. What is the charge of a proton? _____
21. What is the charge of an electron? _____
22. What is the charge of a neutron? _____
23. Why doesn't an electron fall into the nucleus? _____
24. What holds the nucleus together? _____
25. What are the four types of forces inside an atom? _____

Foundations Of Physical Science Answers To Chapter 9

CO Houle



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Table of Contents Foundations Of Physical Science Answers To Chapter 9

1. Understanding the eBook Foundations Of Physical Science Answers To Chapter 9
 - The Rise of Digital Reading Foundations Of Physical Science Answers To Chapter 9
 - Advantages of eBooks Over Traditional Books
2. Identifying Foundations Of Physical Science Answers To Chapter 9
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Foundations Of Physical Science Answers To Chapter 9
 - User-Friendly Interface
4. Exploring eBook Recommendations from Foundations Of Physical Science Answers To Chapter 9
 - Personalized Recommendations
 - Foundations Of Physical Science Answers To Chapter 9 User Reviews and Ratings
 - Foundations Of Physical Science Answers To Chapter 9 and Bestseller Lists
5. Accessing Foundations Of Physical Science Answers To Chapter 9 Free and Paid eBooks
 - Foundations Of Physical Science Answers To Chapter 9 Public Domain eBooks
 - Foundations Of Physical Science Answers To Chapter 9 eBook Subscription Services
 - Foundations Of Physical Science Answers To Chapter 9 Budget-Friendly Options
6. Navigating Foundations Of Physical Science Answers To Chapter 9 eBook Formats

- ePub, PDF, MOBI, and More
- Foundations Of Physical Science Answers To Chapter 9 Compatibility with Devices
- Foundations Of Physical Science Answers To Chapter 9 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Foundations Of Physical Science Answers To Chapter 9
 - Highlighting and Note-Taking Foundations Of Physical Science Answers To Chapter 9
 - Interactive Elements Foundations Of Physical Science Answers To Chapter 9
- 8. Staying Engaged with Foundations Of Physical Science Answers To Chapter 9
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Foundations Of Physical Science Answers To Chapter 9
- 9. Balancing eBooks and Physical Books Foundations Of Physical Science Answers To Chapter 9
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Foundations Of Physical Science Answers To Chapter 9
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Foundations Of Physical Science Answers To Chapter 9
 - Setting Reading Goals Foundations Of Physical Science Answers To Chapter 9
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Foundations Of Physical Science Answers To Chapter 9
 - Fact-Checking eBook Content of Foundations Of Physical Science Answers To Chapter 9
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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