

The Hydrogen Atom – Student Guide

Background Material

Carefully read the background pages created through Eureka, Slides, and Experiments and answer the following questions in detail and with understanding.

Question 1: Complete the following table which compares how the Bohr model and the Quantum model compare the Hydrogen atom. In each case, they both describe things in a unique way and in some ways they do not.

Concept Name	Bohr Model	Quantum Model
Electrons	positively charged particles in the center	positively charged particles in the center
Orbitals	negatively charged and orbit the center	negatively charged clouds
Location of the electron in the time interval	2 in three fixed positions	electron cloud position variable the time interval
Energy of the electron Energy Levels	$E = E_n$	$E = E_n$

Question 2: Complete the following table which shows the comparison of two different theories by putting predictions made by each theory side-by-side. The table can be completed by using the Bohr model or the Quantum model. Significant differences, similar responses, similar energy, and the same results through nature will be shown.

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SB Merriam



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