# Unit 18: Hydrocarbon and its Compounds

#### Things to be Studied in this Unit

- Introduction to hydrocarbon
- Saturated and unsaturated hydrocarbon
- Alkane, alkene and alkyne
- Homologous series
- Alkyl radical
- Functional group
- IUPAC system
- Common system to nomenclature of hydrocarbons
- Some hydrocarbon and its compounds
  - Methane
  - Propane
  - Alcohol

- Ethane
  - Butane Glycerol

#### Introduction

Hydrocarbons are organic compounds composed of hydrogen (H) and carbon (C) atoms. They are of two main types, viz. saturated and Unsaturated. The hydrocarbons in which carbon atoms are combined with a single bond are called statured hydrocarbons. The hydrocarbons which contain a double bond or triple bond between carbon atoms are called unsaturated hydrocarbons. Hydrocarbons are the basis for many organic compounds and are found in various forms in fossil fuels like mineral oil and natural gas. They serve as essential building blocks for a wide range of chemical products and fuels, playing a vital role in industry and energy production.

#### Things to Know

- Hydrocarbon: Hydrocarbons are the compounds formed by the combination of hydrogen and carbon. Examples: Methane (CH<sub>4</sub>), Propane (C<sub>3</sub>H<sub>8</sub>) etc.
- Unsaturated hydrocarbon: The hydrocarbon which contains a double bond or triple bond between carbon atoms is called unsaturated hydrocarbon. The general formula of unsaturated hydrocarbons is C<sub>n</sub>H<sub>2n</sub> (alkene) and C<sub>n</sub>H<sub>2n-2</sub> (alkyne).
- Saturated hydrocarbon: The hydrocarbon in which carbon atoms are combined with a single bond is called statured hydrocarbon. The general formula of saturated hydrocarbon is C<sub>0</sub>H<sub>2n=2</sub>.
- IUPAC: IUPAC stands for International Union of Pure and Applied Chemistry.
- Alkane: An alkane is a saturated hydrocarbon in which carbon atoms are connected by only single covalent bonds. The general formula of alkanes is C<sub>n</sub>H<sub>2n+2</sub> where 'n' is the number of carbon atoms present in one molecule of alkane. Example: Propane, Pentane, etc.
- Alkene: An alkene is a saturated hydrocarbon in which carbon atoms are connected by a double bond. So an alkene contains  $-\stackrel{1}{\mathbf{C}} = \stackrel{1}{\mathbf{C}} -$  group. The general formula of an alkene is  $\mathbf{C}_n\mathbf{H}_{2n}$  where 'n' is the number of carbon atoms in its one molecule. Example: Propene, Pentene, etc.

# **Hydrocarbon Compounds Workbook Answers**

**AN Whitehead** 

**Hydrocarbon Compounds Workbook Answers:** 

Unveiling the Magic of Words: A Report on "Hydrocarbon Compounds Workbook Answers"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Hydrocarbon Compounds Workbook Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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