Molarity can be calculated from grams of solute by converting to moles of solute.

$$molarity = \frac{number of moles of solute}{liter of solution} = \frac{moles}{liter}$$

$$moles solute = \frac{g solute}{molar mass} \qquad Molar mass = \frac{g}{mole}$$

$$Molarity = \frac{(g solute) (1 mole)}{(molar mass g)} = \frac{moles}{liter}$$

$$Liter of solution$$

$$\frac{\text{g solute}}{\text{molar mass x liter of solution}} = \frac{\text{moles}}{\text{liter}}$$

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