

**Potentiometry**

Choose the correct answer:

| No | Sentence                                                                                                                                                                                                                                                                             | answer |
|----|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------|
| 1  | All of the following are related to reference electrode except:<br>a) of known and constant potential<br>b) obeys Nernst equation.<br>c) of high resistance to permit current of electric current.<br>d) insensitive to activity of analyte solution.                                | C      |
| 2  | Potential of standard hydrogen electrode is:<br>a) Zero<br>b) related to $H^+$ concentration<br>c) 0.059 PH<br>d) A and B<br>e) All of the above                                                                                                                                     | d      |
| 3  | The increase in chloride ion concentration in Ag/AgCl electrode leads to:<br>a) increase the potential<br>b) decrease the potential<br>c) no effect<br>d) none of them                                                                                                               | b      |
| 4  | Potential of calomel electrode is ..... proportional to chloride ion concentration.<br>a) directly<br>b) inversely<br>c) equal<br>d) no relation                                                                                                                                     | b      |
| 5  | In silver/silver chloride electrode, as KCl concentration increases:<br>a) silver ions increases, potential increases.<br>b) silver ions decreases, potential increases.<br>c) silver ions decreases, potential remains the same.<br>d) silver ions decreases, potential decreases.  | d      |
| 6  | In calomel electrode, as KCl concentration increases:<br>a) mercury(I) ions increases, potential increases.<br>b) mercury(I) ions decreases, potential increases.<br>c) mercury(I) ions decreases, potential remains the same.<br>d) mercury(I) ions decreases, potential decreases. | d      |
| 7  | What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in $Cd^{2+}$ ( $E^0 = -0.403$ V)<br>a) 0.457<br>b) 0.403<br>c) -0.457<br>d) -0.201                                                                         | C      |

# Mcq On Potentiometry

**Wolfgang Guggemos**



## **Mcq On Potentiometry:**

## The Enigmatic Realm of **Mcq On Potentiometry**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Mcq On Potentiometry** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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