

**EARTH AND LIFE SCIENCE**  
**2<sup>ND</sup> QUARTER**  
**G12 ABM- REVIEWER**

**LESSON 1:**

**Exogenic Processes**

- It is a **blanket term** for those processes which operate on or close to the surface of the Earth.
- It includes **geological phenomena and processes** that originate externally to the Earth's surface.

**Types of Exogenic Processes**

**Weathering**

- It is the **breaking down of rocks** into fragments or into smaller size.
- Every rock had undergone weathering, whether it is **disintegration (physical weathering)** or **decomposition (chemical weathering)**.

**Physical Weathering**

- Refers to the **breakdown of rocks** into pieces without any change in its composition.

**Types of Physical Weathering**

**Unloading**

- It is the **uplifted intrusive igneous rocks**.
- It means rocks from **beneath the surface go above the surface**.
- **Weathering are caused by variety of factors** such as **tectonism and volcanism**.

**Thermal Expansion & Contractions**

- It refers to the **effect of temperature to rocks**.
- When exposed to **high temperature**, rocks **expand**; when exposed to **low temperature**, rocks **contract**.

**Freeze- Thaw Weathering**

- It is **most common in the North pole and South**.
- It occurs when a **fractured rock absorbs water** through its crack then got frozen.
- As mention in Thermal Expansion and contraction, the **colder the temperature**, the molecule in it **expand**.

**Hydration**

- It refers to the **amount of water** being absorbed by the rock.
- *When hydrated – too much or plenty of water is present – the rock softens.*
- *When it is dehydrated – too less or lack of water- the rock hardens.*
- The **repeated process** of this causes weathering

**Salt Crystal Growth**

- It usually **happens in the sea or the seaside**.
- It happens when **salt water comes in contact with rocks**.
- The **wind can also take part in the salt crystal growth**.

**Chemical Weathering**

- Refers to the **changes in the composition of rock**.

**Types of Chemical Weathering**

**Oxidation**

- $O^2$  in  $H^2O$  comes in contact with certain elements in minerals of rocks, and chemical reaction can occur.
- E.g. *Fe and Al -  $Fe^2 O^3$  and  $Al^2 O^3$*

**Hydrolysis**

- $H^2O$  molecules alone or  $CO^2$  in  $H^2O$  react with chemical composition/components of rock minerals to create new components or compounds.
- E.g. *silicate minerals (containing silica and  $O^2$ ) in igneous rock silicate minerals – clay minerals*  
 $= KaSi^2 O^2 + H^2O - Ha/Si$
- Reactions that produces weak acid.

**Solution**

- Rock forming mineral dissolve in water
- E.g. *halite mineral containing rock salt (NaCl) susceptible to solution in  $H^2O$*

**Carbonation**

- Involves  $CO^2$  and  $H^2O$  molecules reacting with, and thereby decomposing rock material.
- E.g. *carbonate rocks ( $CO^2$ ) like limestone (composed of  $CaCO^2$ ) ( $O^2 + H^2O - H^2 CO^2$  (carbonic acid)  $CaCO$  (limestone) +  $H^2 CO^2$  (carbonic acid) +  $Ca HCO^2$ ) calcium bicarbonate*

**Mass Wasting**

- also called mass movement
- **downslope transport of surface materials** in direct response to gravity.
- **heavier objects have greater downward materials** in direct response gravity.
- the **force of gravity encourages rock sediments and soil to move downhill or sloping surfaces**.

# Life Science Question Paper2 G12

**A Gutmann**



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