

Chemistry Midterm Study Guide

Periodic Table of the Elements

Legend:

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Inner Transition Metals
- Lanthanides
- Actinides
- Noble Gases
- Metalloids

Horizontal arrow indicating increasing atomic number from left to right.

- Period** - is horizontally row of elements in the periodic table
ex - Chlorine like sodium and potassium, shares the chemical properties
- Group** - is vertically row of elements in the periodic table
ex - Fluorine like chlorine and bromine (chemical properties) because of electrons in the outer shell
- Electronegativity** - ability to attract electrons (the more electronegative an element is, the more likely it is to attract electrons)
- Ionization Energy** - The ability or tendency of an atom to lose or gain electrons
ex - ionization energy, electronegativity, atomic radius, ionization energy, electronegativity

High School Chemistry Midterm Study Guide

David Baud



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