

3.4

Graph of the change in temperature of a substance as it is heated

(5)



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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Table of Contents Life Science Pquestion Paper Grade2014

1. Understanding the eBook Life Science Pquestion Paper Grade2014
 - The Rise of Digital Reading Life Science Pquestion Paper Grade2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Life Science Pquestion Paper Grade2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Life Science Pquestion Paper Grade2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Life Science Pquestion Paper Grade2014
 - Personalized Recommendations
 - Life Science Pquestion Paper Grade2014 User Reviews and Ratings
 - Life Science Pquestion Paper Grade2014 and Bestseller Lists

5. Accessing Life Science Pquestion Paper Grade2014 Free and Paid eBooks
 - Life Science Pquestion Paper Grade2014 Public Domain eBooks
 - Life Science Pquestion Paper Grade2014 eBook Subscription Services
 - Life Science Pquestion Paper Grade2014 Budget-Friendly Options
6. Navigating Life Science Pquestion Paper Grade2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Life Science Pquestion Paper Grade2014 Compatibility with Devices
 - Life Science Pquestion Paper Grade2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Life Science Pquestion Paper Grade2014
 - Highlighting and Note-Taking Life Science Pquestion Paper Grade2014
 - Interactive Elements Life Science Pquestion Paper Grade2014
8. Staying Engaged with Life Science Pquestion Paper Grade2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Life Science Pquestion Paper Grade2014
9. Balancing eBooks and Physical Books Life Science Pquestion Paper Grade2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Life Science Pquestion Paper Grade2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Life Science Pquestion Paper Grade2014
 - Setting Reading Goals Life Science Pquestion Paper Grade2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Life Science Pquestion Paper Grade2014
 - Fact-Checking eBook Content of Life Science Pquestion Paper Grade2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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