

Interpretation: Writing Binary Formulas

Example: Andrew Morgan Pre-test: 2 October 2020/2021

Using electronegativity as a guide to tell whether the atoms are what anions or cations,
 Sodium reacting with Sulfur Hexafluoride, then the total positive charge is always equal to the
 total negative charge for negative charge balanced. The number of electrons lost for
 cations (equal to the number of electrons gained for anions) is the same. The total
 number of electrons lost, balanced by the number of electrons gained. The charge of the
 molecule is always equal to the number of electrons gained and the number of
 electrons lost. Electronegativity is the ability of an atom to attract electrons to itself.

Electronegativity

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 number of electrons lost, balanced by the number of electrons gained. The charge of the
 molecule is always equal to the number of electrons gained and the number of
 electrons lost. Electronegativity is the ability of an atom to attract electrons to itself.

	Cl	S	F	N	O	P
Mg	12/10	12/10	12/10	12/10	12/10	12/10
Ca	12/10	12/10	12/10	12/10	12/10	12/10
Cu	12/10	12/10	12/10	12/10	12/10	12/10
Na	12/10	12/10	12/10	12/10	12/10	12/10
Zn	12/10	12/10	12/10	12/10	12/10	12/10
Al	12/10	12/10	12/10	12/10	12/10	12/10

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CH Cherryholmes



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charles passes his exams and becomes a doctor