

Electrons particles in the nucleus of atoms.
 1. positive
 2. negative

Atomic Number number of protons in the nucleus of an atom.

Electron shells all of _____ (1) = all of _____ (2)

Isotopes atoms of an element that have different numbers of neutrons.

Isotopes of Hydrogen

Hydrogen-1

1 proton and 0 neutrons

Hydrogen-2

1 proton and 1 neutron

Hydrogen-3

1 proton and 2 neutrons

Mass Number total number of protons and neutrons in an atom.

Example: Carbon-12 12 Nitrogen-14 14

Particle	Charge	Mass	Location
Protons	+1	1 atomic mass unit	Nucleus
Neutrons	0	1 atomic mass unit	Nucleus
Electrons	-1	1/1836 atomic mass unit	Electron cloud

Atomic Mass average of the masses of all the isotopes of an element.

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Gpb Note Taking Guide Answers Episode 302

Richard Bailey



Gpb Note Taking Guide Answers Episode 302:

Unveiling the Magic of Words: A Review of "**Gpb Note Taking Guide Answers Episode 302**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Gpb Note Taking Guide Answers Episode 302**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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