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## NOVA: Hunting the Elements

## Part 1: Basic Chemistry

## Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

|                          |                     |                       |                     |                      |                      |                       |                     |                       |                     |                      |                        |                        |                      |                      |                      |                     |                     |                 |
|--------------------------|---------------------|-----------------------|---------------------|----------------------|----------------------|-----------------------|---------------------|-----------------------|---------------------|----------------------|------------------------|------------------------|----------------------|----------------------|----------------------|---------------------|---------------------|-----------------|
| 1<br>H<br>1.00794        |                     |                       |                     |                      |                      |                       |                     |                       |                     |                      |                        |                        |                      |                      |                      |                     | 2<br>He<br>4.002602 |                 |
| 3<br>Li<br>6.941         | 4<br>Be<br>9.00918  |                       |                     |                      |                      |                       |                     |                       |                     |                      |                        | 5<br>B<br>10.811       | 6<br>C<br>12.011     | 7<br>N<br>14.00643   | 8<br>O<br>15.999     | 9<br>F<br>18.998403 | 10<br>Ne<br>20.1797 |                 |
| 11<br>Na<br>22.98976928  | 12<br>Mg<br>24.304  |                       |                     |                      |                      |                       |                     |                       |                     |                      |                        | 13<br>Al<br>26.9815386 | 14<br>Si<br>28.08558 | 15<br>P<br>30.973762 | 16<br>S<br>32.06     | 17<br>Cl<br>35.453  | 18<br>Ar<br>39.948  |                 |
| 19<br>K<br>39.0983       | 20<br>Ca<br>40.078  | 21<br>Sc<br>44.955912 | 22<br>Ti<br>47.88   | 23<br>V<br>50.9415   | 24<br>Cr<br>51.9961  | 25<br>Mn<br>54.938045 | 26<br>Fe<br>55.845  | 27<br>Co<br>58.933195 | 28<br>Ni<br>58.6934 | 29<br>Cu<br>63.546   | 30<br>Zn<br>65.38      | 31<br>Ga<br>69.723     | 32<br>Ge<br>72.64    | 33<br>As<br>74.9216  | 34<br>Se<br>78.96    | 35<br>Br<br>79.904  | 36<br>Kr<br>83.80   |                 |
| 37<br>Rb<br>85.4678      | 38<br>Sr<br>87.62   | 39<br>Y<br>88.90584   | 40<br>Zr<br>91.224  | 41<br>Nb<br>92.90638 | 42<br>Mo<br>95.94    | 43<br>Tc<br>98.9062   | 44<br>Ru<br>101.07  | 45<br>Rh<br>102.9055  | 46<br>Pd<br>106.42  | 47<br>Ag<br>107.8682 | 48<br>Cd<br>112.411    | 49<br>In<br>114.818    | 50<br>Sn<br>118.710  | 51<br>Sb<br>121.757  | 52<br>Te<br>127.6    | 53<br>I<br>126.905  | 54<br>Xe<br>131.29  |                 |
| 55<br>Cs<br>132.90545196 | 56<br>Ba<br>137.327 | 57-70<br>(*)          | 71<br>Lu<br>174.967 | 72<br>Hf<br>178.49   | 73<br>Ta<br>180.9479 | 74<br>W<br>183.84     | 75<br>Re<br>186.207 | 76<br>Os<br>190.23    | 77<br>Ir<br>192.222 | 78<br>Pt<br>195.084  | 79<br>Au<br>196.966569 | 80<br>Hg<br>200.59     | 81<br>Tl<br>204.3833 | 82<br>Pb<br>207.2    | 83<br>Bi<br>208.9804 | 84<br>Po<br>209     | 85<br>At<br>210     | 86<br>Rn<br>222 |
| 87<br>Fr<br>223          | 88<br>Ra<br>226     | 89-102<br>(*) (*)     | 103<br>Lr<br>260    | 104<br>Rf<br>261     | 105<br>Db<br>262     | 106<br>Sg<br>266      | 107<br>Bh<br>264    | 108<br>Hs<br>277      | 109<br>Mt<br>268    | 110<br>Uun<br>288    | 111<br>Uuu<br>289      | 112<br>Uub<br>288      | 114<br>Uuq<br>289    |                      |                      |                     |                     |                 |

<sup>a</sup> Lanthanide series.

|          |          |          |          |          |          |          |          |          |          |          |           |           |           |
|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|-----------|-----------|-----------|
| 57<br>La | 58<br>Ce | 59<br>Pr | 60<br>Nd | 61<br>Pm | 62<br>Sm | 63<br>Eu | 64<br>Gd | 65<br>Tb | 66<br>Dy | 67<br>Ho | 68<br>Er  | 69<br>Tm  | 70<br>Yb  |
| 89<br>Ac | 90<br>Th | 91<br>Pa | 92<br>U  | 93<br>Np | 94<br>Pu | 95<br>Am | 96<br>Cm | 97<br>Bk | 98<br>Cf | 99<br>Es | 100<br>Fm | 101<br>Md | 102<br>No |

[illegible]

**Gold - Au**

1. Write the number of subatomic particles in gold:
  - a. Protons;
  - b. Neutrons;
  - c. Electrons;
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

# Hunting The Elements Answers

**JA Banks**



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