



Student Exploration: Phase Changes

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: altitude, boil, boiling point, freeze, freezing point, gas, liquid, melt, melting point, phase, solid

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Marla wants a hard-boiled egg for breakfast. The recipe says boil water, then lower the heat so the water is simmering. But Marla is in a rush, so she keeps the heat on high as the egg cooks.

Do you think the egg will be done faster if the heat is kept on high as the water boils?

No

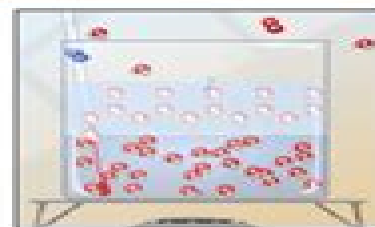
Explain your answer.

If the water is kept high the egg will cook inconsistently so the outer and inner layers will be at different stages.

Gizmo Warm-up

In the *Phase Changes* Gizmo, select **Micro view** and set the **Ice volume** to 50 cm³. Notice the nitrogen (N₂), oxygen (O₂), and water (H₂O) molecules.

To start, create ice. Set **Add/remove heat energy** to -200 J/s, click **Play** (▶), and then click **Pause** (⏸) when the container is about half ice, half water. Set **Add/remove heat energy** back to 0 J/s, and then click **Play**.



1. Observe water molecules in the **solid** (ice), **liquid** (water), and **gas** (air) **phases**.

- A. In which phase(s) do molecules move freely?
- B. In which phase(s) are molecules held together?
- C. What do you notice about the positions of the ice molecules?

Gas

Solid

They are all about equal distance apart and more uniform than the air or water molecules.

2. Click **Reset** (↺), and select **Macro view**. Set the **Ice volume** to 25 cm³.

- A. In which phase(s) is water held in a defined shape?
- B. In which phase(s) does the water take the shape of the container?

Solid

Liquid

Gizmo Student Exploration Phase Changes Answer Key

Rachel Sandford



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Table of Contents Gizmo Student Exploration Phase Changes Answer Key

1. Understanding the eBook Gizmo Student Exploration Phase Changes Answer Key
 - The Rise of Digital Reading Gizmo Student Exploration Phase Changes Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploration Phase Changes Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Student Exploration Phase Changes Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Student Exploration Phase Changes Answer Key
 - Personalized Recommendations
 - Gizmo Student Exploration Phase Changes Answer Key User Reviews and Ratings
 - Gizmo Student Exploration Phase Changes Answer Key and Bestseller Lists

5. Accessing Gizmo Student Exploration Phase Changes Answer Key Free and Paid eBooks
 - Gizmo Student Exploration Phase Changes Answer Key Public Domain eBooks
 - Gizmo Student Exploration Phase Changes Answer Key eBook Subscription Services
 - Gizmo Student Exploration Phase Changes Answer Key Budget-Friendly Options
6. Navigating Gizmo Student Exploration Phase Changes Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Student Exploration Phase Changes Answer Key Compatibility with Devices
 - Gizmo Student Exploration Phase Changes Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploration Phase Changes Answer Key
 - Highlighting and Note-Taking Gizmo Student Exploration Phase Changes Answer Key
 - Interactive Elements Gizmo Student Exploration Phase Changes Answer Key
8. Staying Engaged with Gizmo Student Exploration Phase Changes Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploration Phase Changes Answer Key
9. Balancing eBooks and Physical Books Gizmo Student Exploration Phase Changes Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploration Phase Changes Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Student Exploration Phase Changes Answer Key
 - Setting Reading Goals Gizmo Student Exploration Phase Changes Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Student Exploration Phase Changes Answer Key
 - Fact-Checking eBook Content of Gizmo Student Exploration Phase Changes Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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