Name:

Tra Nguyen

Date:

11/12/2020

#### Student Exploration: Electron Configuration

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."

Elvis Perkins will most likely sit on the seat on the upper left that is empty with no passengers.

Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."

MArtta Warren will most likely sit next to the door with the white an black fellow. Since she is fired and will want to not have to walk all the way down the rows when it is her time to exit.



In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? \_\_\_\_\_

No, people tend to sit and give others space if there is an empty seat available on a bus.

#### Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration* Gizmo.

To begin, check that Lithium is selected on the PERIODIC TABLE tab.

1. The atomic number is equal to the number of protons in an atom,

How many protons are in a lithium atom? \_\_\_\_\_

There are 7 protons in a Lithium atom.





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